

# Physics Formula Sheet for NEET Exam

Some of the topicwise Physics formulas, which can solve questions directly in NEET exam, have been provided below:

## Kinematics Formula

### Average and Instantaneous Vel. and Accel.:

$$\vec{v}_{av} = \Delta \vec{r} / \Delta t,$$

$$\vec{v}_{inst} = d\vec{r} / dt$$

$$\vec{a}_{av} = \Delta \vec{v} / \Delta t$$

$$\vec{a}_{inst} = d\vec{v} / dt$$

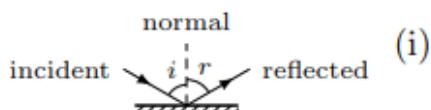
### Motion in a straight line with constant $a$ :

$$v = u + at, \quad s = ut + \frac{1}{2}at^2, \quad v^2 - u^2 = 2as$$

**Relative Velocity:**  $\vec{v}_{A/B} = \vec{v}_A - \vec{v}_B$

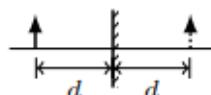
## Reflection of light

### Laws of reflection:



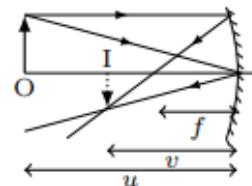
Incident ray, reflected ray, and normal lie in the same plane (ii)  $\angle i = \angle r$

### Plane mirror:



(i) the image and the object are equidistant from mirror (ii) virtual image of real object

### Spherical Mirror:



1. Focal length  $f = R/2$
2. Mirror equation:  $\frac{1}{v} + \frac{1}{u} = \frac{1}{f}$
3. Magnification:  $m = -\frac{v}{u}$

## Specific Heat Capacity

**Specific heat:**  $s = \frac{Q}{m\Delta T}$

**Latent heat:**  $L = Q/m$

**Specific heat at constant volume:**  $C_v = \left. \frac{\Delta Q}{n\Delta T} \right|_V$

**Specific heat at constant pressure:**  $C_p = \left. \frac{\Delta Q}{n\Delta T} \right|_p$

**Relation between  $C_p$  and  $C_v$ :**  $C_p - C_v = R$

**Ratio of specific heats:**  $\gamma = C_p/C_v$

**Relation between  $U$  and  $C_v$ :**  $\Delta U = nC_v\Delta T$

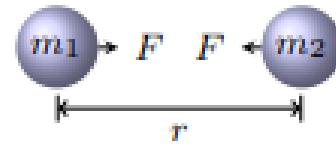
**Specific heat of gas mixture:**

$$C_v = \frac{n_1 C_{v1} + n_2 C_{v2}}{n_1 + n_2}, \quad \gamma = \frac{n_1 C_{p1} + n_2 C_{p2}}{n_1 C_{v1} + n_2 C_{v2}}$$

**Molar internal energy of an ideal gas:**  $U = \frac{f}{2}RT$ ,  
 $f = 3$  for monatomic and  $f = 5$  for diatomic gas.

# Gravitation

**Gravitational force:**  $F = G \frac{m_1 m_2}{r^2}$



**Potential energy:**  $U = -\frac{GMm}{r}$

**Gravitational acceleration:**  $g = \frac{GM}{R^2}$

**Variation of g with depth:**  $g_{\text{inside}} \approx g \left(1 - \frac{2h}{R}\right)$

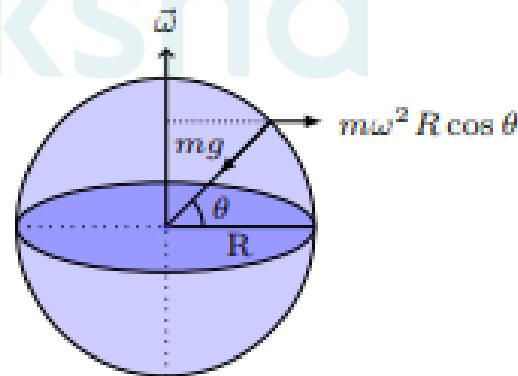
**Variation of g with height:**  $g_{\text{outside}} \approx g \left(1 - \frac{h}{R}\right)$

**Effect of non-spherical earth shape on g:**

$g_{\text{at pole}} > g_{\text{at equator}} \quad (\because R_e - R_p \approx 21 \text{ km})$

**Effect of earth rotation on apparent weight:**

$$mg'_\theta = mg - m\omega^2 R \cos^2 \theta$$



**Orbital velocity of satellite:**  $v_o = \sqrt{\frac{GM}{R}}$

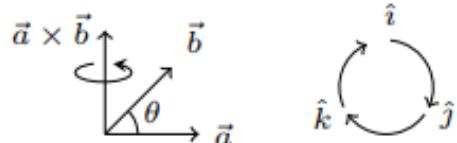
# Vectors

**Notation:**  $\vec{a} = a_x \hat{i} + a_y \hat{j} + a_z \hat{k}$

**Magnitude:**  $a = |\vec{a}| = \sqrt{a_x^2 + a_y^2 + a_z^2}$

**Dot product:**  $\vec{a} \cdot \vec{b} = a_x b_x + a_y b_y + a_z b_z = ab \cos \theta$

**Cross product:**

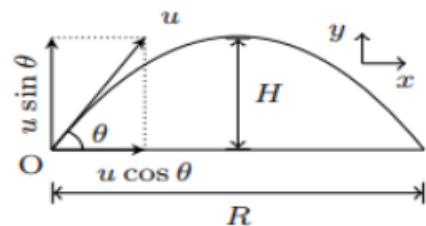


$$\vec{a} \times \vec{b} = (a_y b_z - a_z b_y) \hat{i} + (a_z b_x - a_x b_z) \hat{j} + (a_x b_y - a_y b_x) \hat{k}$$

$$|\vec{a} \times \vec{b}| = ab \sin \theta$$

## Projectile Motion

**Projectile Motion:**



$$x = ut \cos \theta, \quad y = ut \sin \theta - \frac{1}{2}gt^2$$

$$y = x \tan \theta - \frac{g}{2u^2 \cos^2 \theta} x^2$$

$$T = \frac{2u \sin \theta}{g}, \quad R = \frac{u^2 \sin 2\theta}{g}, \quad H = \frac{u^2 \sin^2 \theta}{2g}$$

# Work, Power and Energy

**Work:**  $W = \vec{F} \cdot \vec{S} = FS \cos \theta, \quad W = \int \vec{F} \cdot d\vec{S}$

**Kinetic energy:**  $K = \frac{1}{2}mv^2 = \frac{p^2}{2m}$

**Potential energy:**  $F = -\partial U / \partial x$  for conservative forces.

$$U_{\text{gravitational}} = mgh, \quad U_{\text{spring}} = \frac{1}{2}kx^2$$

**Work done by conservative forces** is path independent and depends only on initial and final points:  
 $\oint \vec{F}_{\text{conservative}} \cdot d\vec{r} = 0$ .

**Work-energy theorem:**  $W = \Delta K$

**Mechanical energy:**  $E = U + K$ . Conserved if forces are conservative in nature.

**Power**  $P_{\text{av}} = \frac{\Delta W}{\Delta t}, \quad P_{\text{inst}} = \vec{F} \cdot \vec{v}$

## Heat and Temperature

**Temp. scales:**  $F = 32 + \frac{9}{5}C, \quad K = C + 273.16$

**Ideal gas equation:**  $pV = nRT, \quad n$  : number of moles

**van der Waals equation:**  $(p + \frac{a}{V^2})(V - b) = nRT$

**Thermal expansion:**  $L = L_0(1 + \alpha \Delta T),$   
 $A = A_0(1 + \beta \Delta T), \quad V = V_0(1 + \gamma \Delta T), \quad \gamma = 2\beta = 3\alpha$

**Thermal stress of a material:**  $\frac{F}{A} = Y \frac{\Delta l}{l}$

## Photoelectric effect

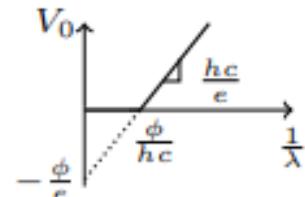
**Photon's energy:**  $E = h\nu = hc/\lambda$

**Photon's momentum:**  $p = h/\lambda = E/c$

**Max. KE of ejected photo-electron:**  $K_{\max} = h\nu - \phi$

**Threshold freq. in photo-electric effect:**  $\nu_0 = \phi/h$

**Stopping potential:**  $V_o = \frac{hc}{e} \left( \frac{1}{\lambda} \right) - \frac{\phi}{e}$



**de Broglie wavelength:**  $\lambda = h/p$