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The number of elements in the relation
 $R = \{(x, y) : 4x^2 + y^2 < 52, x, y \in \mathbb{Z}\}$ is

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If the mean deviation about the median of the numbers $k, 2k, 3k \dots 1000k$ is 500, then k^2 is equal to

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Let $s = \{z \in \mathbb{C} : 4z^2 + \bar{z} = 0\}$. Then $\sum_{z \in s} |z|^2$ is equal to.

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In an open organ pipe f_3 and f_6 are 3rd and 6th harmonic frequencies respectively and if $f_6 - f_3 = 2200\text{Hz}$.

Then the length of pipe is (in mm)

- a. 225
- b. 200
- c. 250
- d. 275

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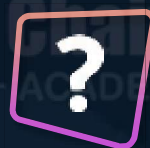
3 small identical bubbles of water having same charge on each coalesce to form a bigger bubble, Then the ratio of the potentials on one initial bubble & that on the resultant bigger bubble is:

(a) $1:3^{2/3}$

(b) $3^{2/3}:1$

(c) $1:2^{2/3}$

(d) $1:3^{1/3}$.



Five positive charges each having charge q are placed at the vertices of a pentagon as shown in the figure. The electric potential (V) & the electric field (\vec{E}) at the center O of the pentagon due to the 5 positive charges are:-

a) $V = 0, E = 0$

b)

$$V = \frac{5q}{4\pi\epsilon_0 r}$$

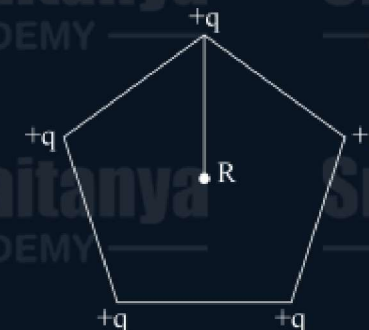
$$E = \frac{5q}{4\pi\epsilon_0 r^2}$$

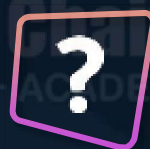
c) $V = \frac{5a}{4\pi\epsilon_0 r} \vec{E} = 0$

d)

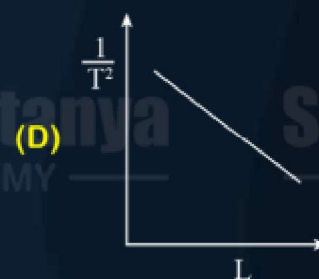
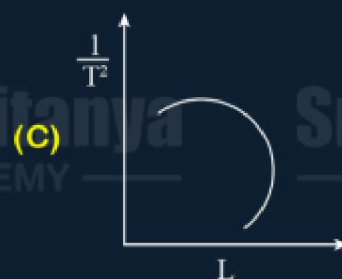
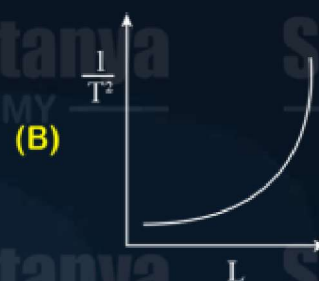
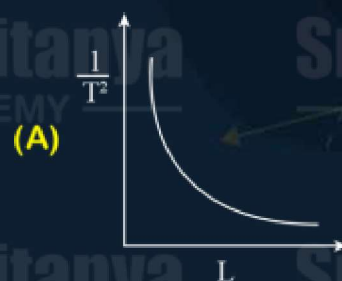
$$V = \frac{5q}{4\pi\epsilon_0 r}$$

$$E = \frac{5\sqrt{3}q}{8\pi\epsilon_0 r^2} \hat{r}$$





using a simple pendulum experiment g is determined by measuring its time period T . Which of the following plots represent correct relation b/w the pendulum length ℓ & time period T .



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Which of the following mixture gives a buffer solution with $\text{pH}=9.5$? Given $\text{PK}_b(\text{NH}_4\text{OH})=4.75$

- a) $0.2\text{M NH}_4\text{OH}(0.4\text{l}) + 0.1\text{M HCl}(1\text{l})$
- b) $0.4\text{M NH}_4\text{OH}(1\text{l}) + 0.1\text{M HCl}(1\text{l})$
- c) $0.5\text{M NH}_4\text{OH}(0.2\text{l}) + 0.2\text{M HCl}(0.5\text{l})$
- d) $0.2\text{M NH}_4\text{OH}(0.5\text{l}) + 0.1\text{M HCl}(0.5\text{l})$

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S-I: $C < O < N < F$ is the correct order in terms of first ionization enthalpy values

S-II: $S > Se > Te > Po > O$ is the correct order in terms of the magnitude of electron gain enthalpy values.

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When 1g of compound (X) is subjected to Kjeldahl's method for estimation of nitrogen, 15 mL 1M H_2SO_4 was neutralized by ammonia evolved. The % of nitrogen in compound (X) is

- a) 21
- b) 0.21
- c) 42
- d) 0.42

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Name of IUPAC of the following



1. 2-Bromo-5-methyl propanoate
2. n-propyl-1-bromo-4methyl hexanoate
3. 2-Bromo-5-methyl hexyl propanoate
4. n-propyl-2-bromo-5-methyl heptanoate

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The dibromo compound [P] of molecular formula ($C_9H_{10}Br_2$) when heated with excess Sodamide followed by treatment with dilute HCl gives [Q]. On warming [Q] with mercuric sulphate dilute sulphuric acid yield (R) which gives positive iodoform test but negative tollen's test. The compound [P] is

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100 g of 98% H_2SO_4 of aqueous solution is mixed with 100 g of 49% H_2SO_4 of aqueous solution. The mole fraction of the H_2SO_4 is.

- a) 0.667
- b) 0.336
- c) 0.1
- d) 0.9

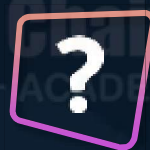
? If $\lim_{x \rightarrow 0} \frac{e^{(a-1)x} + 2\cos bx + (c-2)e^{-x}}{x \cos x - \log_e(1+x)} = 2$, then $a^2 + b^2 + c^2$ is equal to?

?

Let the domain of function

$$f(x) = \log_3 \log_5 (7 - \log_2 (x^2 - 10x + 85)) + \sin^{-1} \left(\left| \frac{3x-7}{17-x} \right| \right) \text{ be } (\alpha, \beta)$$

Then $\alpha + \beta$ is equal



$\cos(\alpha + \beta) = -\frac{1}{10}$ and $\sin(\alpha - \beta) = \frac{3}{8}$ where $0 < \alpha < \frac{\pi}{3}$ & $0 < \beta < \frac{\pi}{4}$
if $\tan 2\alpha = \frac{3(1-\gamma\sqrt{5})}{\sqrt{11}(s+\sqrt{5})}$, $\gamma, s \in N$, then $r + s$ is equal to ____

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The area of the region $A = \{(x, y) : 4x^2 + y^2 \leq 8 \text{ and } y^2 \leq 4x\}$ is

- A) $\pi/2 + 2$
- B) $\frac{\pi}{2} + \frac{1}{3}$
- C) $\pi + \frac{2}{3}$
- D) $\pi + 4$

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let $P(10, 2\sqrt{15})$ be a point on the hyperbola $\frac{x^2}{a^2} - \frac{y^2}{b^2} = 1$ whose foci are S and S' . If the length of its latus rectum is 8 then the square of the area of $\triangle PSS'$ is equal to

- A) 900
- B) 4200
- C) 1462
- D) 2700 .

