

JEE Mains 2026 Session 1



DETAILED  **LIVE**
PAPER SOLUTIONS

28th Jan - Shift 1



Let $f(x)$ be a polynomial function such that

$$f(x^2 + 1) = x^4 + 5x^2 + 3, \text{ then } \int_0^3 f(x) =$$

?

If unit vectors $\vec{a}, \vec{b}, \vec{c}$ then $|\vec{a} - \vec{b}|^2 + |\vec{b} - \vec{c}|^2 + |\vec{c} - \vec{a}|^2 = 9$, $|2\vec{a} + k\vec{b} + kc| = 3$, then find the positive values of k .



If α, β , where $\alpha < \beta$ are the roots of the equation $\lambda x^2 - (\lambda + 3)x + 3 = 0$ and $\frac{1}{\alpha} - \frac{1}{\beta} = \frac{1}{3}$, then the sum of all possible values of λ is

- A) 8
- B) 4
- C) 2
- D) 6



If $g(x) = 3x^2 + 2x - 3$, $f(0) = -3$, $4g(f(x)) = 3x^2 - 32x + 72$. Then $f(g(2))$ is equal to

- A) $-\frac{25}{6}$
- B) $\frac{25}{6}$
- C) $-\frac{7}{2}$
- D) $\frac{7}{2}$

?

Consider the 10 observations 2, 3, 5, 10, 11, 13, 15, 21, a and b such that mean of observation is 9 and variance is 34.2. Then the mean deviation about median is?

- A) 3
- B) 5
- C) 6
- D) 7

?

In Carius method of estimation of 'Br', 1.53 g of an organic compound gave 1 g of AgBr. The percentage of Br in organic compound is _____

(Atomic mass of Ag & Br is 108 & 80u respectively)

1) 35.23

2) 43.53

3) 27.81

4) 22.71

?

In period 4 of the periodic table which elements have the highest and lowest atomic radii respectively

1) K and Br

2) Na and Cl

3) K and Se

4) Rb and Br

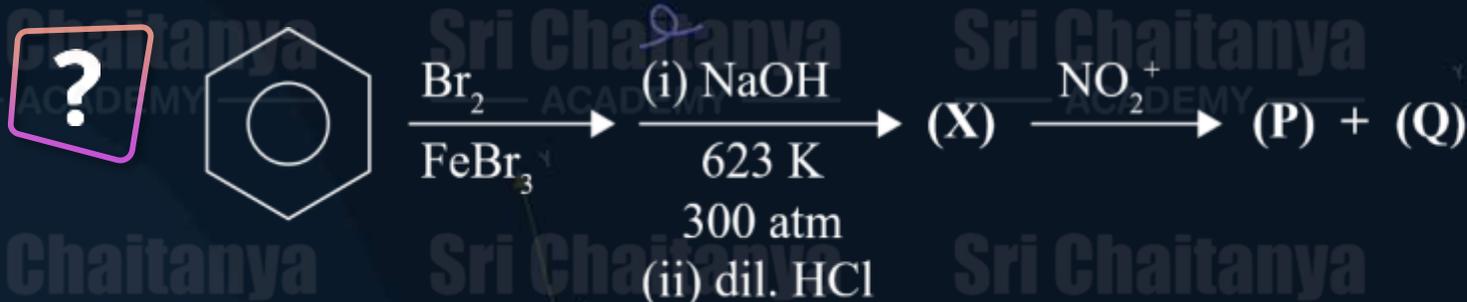


Consider the following nickel complexes:



Which of the following options correctly describes the magnetic behaviour (paramagnetic/diamagnetic) of these complexes?

- 1) A, B are diamagnetic; C is paramagnetic
- 2) A, B are paramagnetic; C is diamagnetic
- 3) A, C are diamagnetic; B is paramagnetic
- 4) A, C are paramagnetic; B is diamagnetic

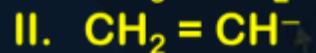
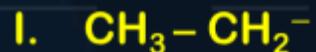


'P' and 'Q' can be separated by

- 1) Simple distillation
- 2) Fractional distillation
- 3) Steam distillation
- 4) Sublimation

?

Consider Following ions



Stability of ions is in the order





Lens of focal length $f = 18\text{cm}$ has refractive index $3/2$. It is immersed in water of refractive index $4/3$. Change in focal length is $\alpha \times f$, value of α is _____

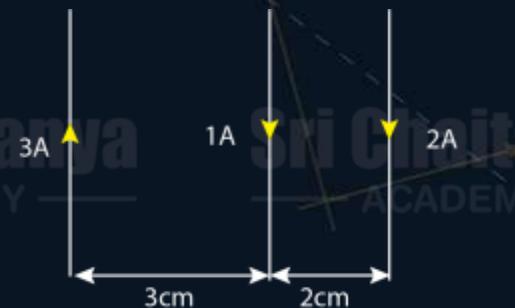


A body of mass 5kg is placed on a rough inclined plane of angle 30 degrees and coefficient of friction $\sqrt{3}/2$. Find the force required to push the body down at constant velocity.



There are three long parallel wires in a plane as shown. Find force on 15 cm of length of middle wire.

- a. $5\mu\text{N}$
- b. $7\mu\text{N}$
- c. $6\mu\text{N}$
- d. μN





Solid sphere with radius 10 cm rotating about axis which is at 15 cm from COM of sphere - Radius of gyration

is \sqrt{n} cm. The value of $n = ?$



Equation of an EMW in a medium is given by

$$E = 2\sin(2 \times 10^{15}t - 10^7x). \text{ Find refractive index of the medium.}$$

a. $3/2$

b. 2

c. $5/3$

d. $4/3$



For a circular coil of radius R , magnetic field at the center of circular coil is $B_0 = 16\mu\text{ T}$. What will be the magnetic field on axis

at a distance $x = \sqrt{3}R$ from center?

a. $\frac{1}{4}\mu\text{ T}$

b. $\frac{1}{2}\mu\text{ T}$

c. $4\mu\text{ T}$

d. $2\mu\text{ T}$



Let $\tan(\pi/4 + 1/2\cos^{-1}2/3) + \tan^{-1}(1/2\sin^{-1}2/3) = k$. Then number of solutions of the equation $\sin^{-1}(kx - 1) = \sin^{-1}x - \cos^{-1}x$ is

?

The value of $\sum_{k=1}^{\infty} (-1)^{k+1} \left(\frac{k(k+1)}{k!} \right)$ is

- A) $e/2$
- B) $2/e$
- C) \sqrt{e}
- D) $1/e$



$$\lim_{x \rightarrow 0} \frac{\ln(\sec(ex)\sec(e^2x)\sec(e^3x) \dots \sec(e^{10}x))}{e^2 - e^2 \cos x}$$



$$\int \frac{1-5\cos^2 x}{\sin^5 x \cos^2 x} dx = f(x) + c, \text{ then } f\left(\frac{\pi}{4}\right) - f\left(\frac{\pi}{6}\right) =$$

?

Let $S = \{1,2,3,4,5,6,7,8,9\}$. Let x be the number of 9-digit number formed using the digits of the set S such that, only one digit is repeated and it is repeated exactly twice. Let ' y ' be the number of 9-digit number formed using the digits of the set S such that, only two digits are repeated and each of these is repeated exactly twice. Then,

A) $56x = 9y$

B) $9x = 2y$

C) $21x = 4y$

D) $45x = 7y$



Let $S = \{x^3 + ax^2 + bx + c; a, b, c \in N \text{ \& } a, b, c \leq 20\}$ be a set of polynomials. Then the number of polynomials in S , which are divisible by $x^2 + 2$, is

- A) 20
- B) 10
- C) 6
- D) 120

?

Let $\tan(\pi/4 + 1/2\cos^{-1}2/3) + \tan^{-1}(1/2\sin^{-1}2/3) = k$. Then number of solutions of the equation $\sin^{-1}(kx - 1) = \sin^{-1}x - \cos^{-1}x$ is



The value of $\sum_{k=1}^{\infty} (-1)^{k+1} \left(\frac{k(k+1)}{k!} \right)$ is

- A) $e/2$
- B) $2/e$
- C) \sqrt{e}
- D) $1/e$

?

The area of region $R = \{(x, y): xy \leq 8, 1 \leq y < x^2, x < 0\}$ is





Which of the following statements is incorrect?

Options:

- 1) P is more acidic than Q
- 2) Q is more acidic than P
- 3) Q is soluble in NaHCO_3
- 4) P and Q both are soluble in NaOH



An organic compound is given. It undergoes 1st order decomposition. It decomposes to 1/8 and 1/10 in time

$t(1/8)$ and $t(1/10)$ respectively. Find out $\frac{t_{1/8}}{t_{1/10}} \times 10 = \underline{\hspace{2cm}}$.

(Take $\log 8 = 0.90$, $\log 7 = 0.84$, $\log 9 = 0.95$)



Given below are two statements.

Statement I : Given the molecules XeF_4 , SiF_4 , SeF_4 and BF_4^- , all the compounds have two different E – F bond lengths, where E is the central atom.

Statement II : Among the species O_2^+ , O_2 , O_2^- and F_2 , the species O_2^- has the highest bond order.

In the light of the above statements, which is the correct option.

- Both statement-I and statement-II are correct
- Both statement-I and statement-II are incorrect
- Statement-I is correct and statement-II is incorrect
- Statement-I is incorrect and statement-II is correct



For equivalence point X ml of 0.02 M HCl is treated with 5 mL of 0.02 M of a weak base. The pK_b of weak base is 5.69 and the pH of the resulting solution is Y at half of the equivalence point. The value of $(x + y)$ is:

- 1) 15
- 2) 8.81
- 3) 13.31
- 4) 3.81

?

Choose the correct statements in respect of hydrides of Group-15.

- A. Reducing power increasing down the group.
- B. Basic nature increases down the group.
- C. Stability decreases down the group.
- D. Boiling point decreases regularly down the group.

- 1) A, B and C only
- 2) A, B and D only
- 3) A and C only
- 4) B, C and D only

?

The wave number of three spectral lines of H -atom are given. Identify the correct set of spectral lines belonging to Balmer series

1) $\frac{5R}{36}, \frac{3R}{16}, \frac{21R}{100}$

2) $\frac{3R}{4}, \frac{3R}{16}, \frac{7R}{144}$

3) $\frac{7R}{144}, \frac{3R}{16}, \frac{16R}{255}$

4) $\frac{5R}{36}, \frac{3R}{16}, \frac{21R}{24}$